CYCLOALKANECARBOXYLIC ACID SYNTHESIS BY INTRAMOLECULAR CHLORO OLEFIN ANNELATION

by Peter T. Lansbury and Robert C. Stewart
Department of Chemistry
State University of New York at Buffalo
Buffalo, New York 14214

(Received in USA 2 January 1973; received in UK for publication 20 March 1973)

Chloro olefin annelation has been shown to have great utility in the construction of cycloalkanone and acylcycloalkane rings, using terminal and internal chloroalkene moleties, respectively, as sites for electrophilic closure. In addition, we have investigated substrates bearing 1,1-dichloro-1-ene side chains, since cationic attack can lead directly to cycloalkanecarboxylic acids², e.g.

This note reports typical five- and six-membered ring syntheses which culminate in closure of the above type, and which illustrate the use of two easily preparable and generally useful alkylating agents

The condensation product $\underline{1}$, derived from chloral and malonic acid³, when successively acetylated, treated with zinc to effect dechloroacetoxylation ($\underline{2}$ - $\underline{3}$), and reduced with lithium aluminum hydride, all without isolation of intermediates provides pure 4,4-dichloro-3-buten-1-ol⁴ ($\underline{4}$), bp 80-81.5°/10mm, in ca. 30% yield (not optimized).

Conversion of $\underline{4}$ to its bromide⁵ $\underline{5}$ and mesylate⁵ $\underline{6}$ by phosphorous tribromide and methanesulfonyl chloride, respectively, allows introduction of the annelating agent into a bifunctional substrate such as desoxybenzoin either electrophilically $(-\underline{8})$.

when ketone ⁵ 7 is reduced and the resulting carbinol ⁵ refluxed overnight in 97% formic acid, cyclization occurs in the same manner as with 8 to afford trans, trans-2,3-diphenylcyclopentanecarboxylic acid (9), mp 150.5-152°, in ca. 50% yield by either route. Thus, prior carbonium ion rearrangement before the non-nucleophilic dichlorobutenyl side chain cyclizes provides a single product as planned. The trans, trans configuration of 9 is assigned by analogy with a similar cyclization which produced trans-3,4-diphenylcyclopentanone and by the absence of epimerization on heating ⁷, which verifies the trans relationships of the

carboxyl group and adjacent phenyl ring.

Six-membered ring formation is exemplified by the sequence below, commencing with the alkylation of cyclohexanone pyrrolidine enamine by 1,1-dichloro-3-bromo-propene⁵. The latter was readily acquired by benzoyl peroxide-initiated N-bromo-succinimide reaction with 1,1-dichloropropene in carbon tetrachloride, affording 66% yield of the expected bromide, bp 65-66.5/17mm. Ketone 10 (semicarbazone⁴: mp 169-170°) was reacted with ethereal vinylmagnesium bromide and the resultant carbinol⁵ formolyzed.

The isomeric octalin carboxylic acids⁵ (11), formed in ca. 40% yield, were hydrogenated (Pt catalyst in acetic acid) to a mixture of <u>cis,cis</u>- and <u>cis,trans</u>-decahydro-2-naphthoic acids, whose identities were established by comparison with authentic samples⁸.

Not surprisingly, these "intramolecular Bott reactions" proceed in more satisfactory yields than were noted in <u>intermolecular</u> combination of hindered 1-methylcycloalkyl cations with 1,1-dichloroethylene² and may have utility in generating the carbocyclic framework of terpenes and of picrotaxane alkaloids such as dendrobine⁹. We also wish to point out that the three- and four-carbon annelating agents used to assemble <u>7</u>, <u>8</u>, and <u>11</u> can be easily homologated and thus extend the possibilities for cycloalkanecarboxylic acid synthesis to rings larger than six-membered.

References

- 1. P. T. Lansbury, Accts. Chem. Res. 5, 311 (1972).
- K. Bott and H. Hellmann, "Newer Methods of Preparative Organic Chemistry", W. Foerst, Ed., Academic Press, New York, N.Y., Vol. 6, 1971, pp. 67-81. This article describes the bimolecular reaction of carbonium ions with 1,1-dichloroethylene in sulfuric acid, leading to substituted acetic acids.
- C. D. Hurd and S. Hayao, J. Amer. Chem. Soc., 76, 5564 (1954).
- Satisfactory elemental analyses, in addition to concordant spectral data, were obtained for this compound.

1572 No. 17

 Characterized by an appropriate combination of ir, nmr and mass spectral data.

- P. T. Lansbury, E. J. Nienhouse, D. J. Scharf and F. R. Hilfiker, J. Amer. Chem. Soc., 92, 5649 (1970).
- 7. H. E. Zimmerman and H. Giallombardo, J. Amer. Chem. Soc., 78, 6259 (1956).
- 8. W. G. Dauben and E. Hoerger, J. Amer. Chem. Soc., 73, 1504 (1951).
- 9. Y. Inubushi, T. Kikuchi, T. Ibuka, K. Tanaka, I. Sajiand and K. Tokane, J. Chem. Soc., Chem. Commun., 1252 (1972).

<u>Acknowledgment</u>: We are grateful to the National Science Foundation for financial support of this research.